## Topic 2.1 How and why do we study matter?

### Pure Substances and Mixtures, pages 45-46

**1.** Pure substances: copper, table salt, water, baking soda, silver, gold, carbon dioxide Mixtures: orange juice, granola cereal, milk, toothpaste, mouthwash, soap, pepperoni pizza, fruit salad, pop, air



3.

|    | Sample             | Type of Matter        | Substances in Mixture       |
|----|--------------------|-----------------------|-----------------------------|
| a) | pulpy orange juice | heterogeneous mixture | water, sugar, orange        |
| b) | granola cereal     | heterogeneous mixture | cereal, nuts, fruits        |
| c) | milk               | homogeneous mixture   | water, sugar, fat           |
| d) | copper             | element               |                             |
| e) | table salt         | compound              |                             |
| f) | baking soda        | compound              |                             |
| g) | water              | compound              |                             |
| h) | toothpaste         | homogeneous mixture   | baking soda, fluoride, gels |
| i) | mouthwash          | homogeneous mixture   | alcohol, water              |
| j) | soap               | homogeneous mixture   | oils, lye                   |
| k) | silver             | element               |                             |
| I) | gold               | element               |                             |
| m) | pepperoni pizza    | heterogeneous mixture | flour, cheese, pepperoni    |
| n) | fruit salad        | heterogeneous mixture | different fruits            |
| o) | рор                | homogeneous mixture   | water, carbon dioxide       |
| p) | air                | homogeneous mixture   | oxygen, nitrogen            |
| q) | carbon dioxide     | compound              |                             |

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### Physical and Chemical Properties, page 47

- **1.** solubility
- 2. colour
- 3. melting point
- 4. texture
- **5.** combustibility
- 6. ability to conduct heat and electricity
- 7. boiling point
- 8. viscosity
- 9. malleability and hardness
- **10.** reactivity with acids
- 11. reactivity with oxygen
- 12. state of matter
- 13. lack of reactivity

## **Observing Chemical Reactions, page 48**

- 1. release of heat and light
- 2. change in odour
- 3. release of heat and light
- 4. release of heat and light
- 5. release of heat and light, formation of a new substance
- 6. change in colour
- 7. formation of gas bubbles
- 8. change in colour
- 9. formation of a new substance
- 10. formation of a precipitate

# Safety First, page 49

|    | Safety Icon             | Safety Hazard   | Precaution  |
|----|-------------------------|---|---|
| a) | Eye Safety              | <ul> <li>chemicals can splash into eyes</li> <li>glass can shatter and fly into eyes</li> </ul>   | Always wear safety goggles when<br>working with chemicals and<br>glassware.   |
| b) | Clothing Protection ැහි | <ul> <li>spills and splatters from harmful chemicals</li> </ul>   | Wear a lab apron to protect his clothes and skin from spills.   |
| c) | Thermal Safety          | <ul> <li>glassware and equipment can get hot and<br/>cause severe burns</li> </ul>  | Handle hot objects carefully by wearing protective mitts.   |
| d) | Skin Protection         | <ul> <li>some chemicals can cause burns or<br/>irritation when skin comes into contact<br/>with them</li> </ul>   | Wear protective gloves to protect<br>himself from chemicals that might<br>burn his skin or cause irritation.  |
| e) | Electrical              | <ul> <li>electrical equipment can cause<br/>electrocution if it comes into contact with<br/>water</li> <li>electrical cords could cause people to trip</li> </ul> | Make sure that hands are dry when<br>touching electrical cords, plugs, and<br>sockets.<br>Place electrical cords where they are<br>visible so people do not trip. |
| f) | Chemical Safety         | <ul> <li>corrosive chemicals can cause severe<br/>burns.</li> <li>chemicals can be toxic</li> </ul>   | Be careful when working with<br>corrosive and toxic chemicals. If he<br>spills some chemical he should wash<br>the area thoroughly.                               |

## 2.1 Assessment, pages 50–54

| <b>1.</b> A  | <b>14.</b> D |
|--------------|--------------|
| <b>2.</b> E  | <b>15.</b> A |
| <b>3.</b> H  | <b>16.</b> C |
| <b>4.</b> C  | <b>17.</b> A |
| <b>5.</b> B  | <b>18.</b> A |
| <b>6.</b> D  | <b>19.</b> A |
| <b>7.</b> G  | <b>20.</b> D |
| <b>8.</b> F  | <b>21.</b> A |
| <b>9.</b> B  | <b>22.</b> A |
| <b>10.</b> C | <b>23.</b> A |
| 11. A        | <b>24.</b> A |
| <b>12.</b> B | <b>25.</b> B |
| <b>13.</b> D | <b>26.</b> C |

Answers

Topic 2.1



28.

