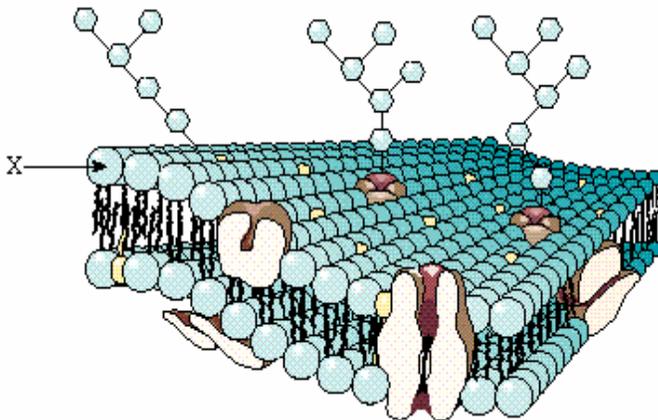


## BIOLOGY 12 - Cell Membrane and Cell Wall Function: Chapter Notes

- The cell membrane is the **gateway into the cell**, and must allow needed things such as **nutrients** into the cell without letting them escape. In the same way, it must allow **wastes** to leave the cell. A **wide variety** of molecules and substances must pass through the cell membrane -- large, small, **hydrophobic, hydrophilic**. Molecules of the **same size** must be sorted out, and the cell must also be able to get **large amounts** of molecules in and out when necessary. How can the cell membrane accomplish this?
- The answer lies in its structure. We already know about the **FLUID MOSAIC MODEL** of membrane structure. Why is it given that name?

### Review of FLUID MOSAIC MODEL:



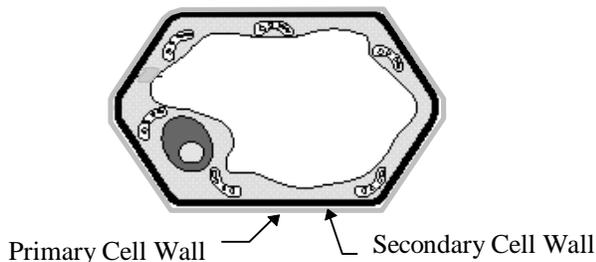
Please Label the Parts of the Cell Membrane

- **double layer** of **phospholipid** molecules ("X" on diagram)
- **consistency of light machine oil (~fluid)**
- **proteins** wholly or partly **embedded** in phospholipid bilayer → forms **mosaic** pattern
- **carbohydrates** strung together in chains are attached to proteins ("**glycoproteins**") or lipids ("**glycolipids**") of membrane. Function as **identification markers** for cell recognition (helps immune

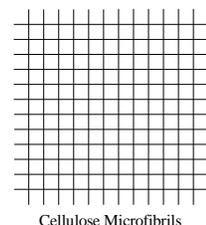
system identify which cells belong to body and which are invaders).

- is **SELECTIVELY PERMEABLE**: some molecules enter the cell, while other molecules (which can be the same size) are not allowed to enter. The cell membrane can **discriminate** between different molecules that are the **same size!**
- all living cells, whether plant, animal, fungal, protozoan, or bacterial, are surrounded by cell membranes

### Plant Cell Wall



- **plants** have **cell walls** in addition to cell membranes. The cell wall lies **outside** the cell membrane.
- (bacteria have cell walls too, but bacterial cell walls are NOT the same in composition as plant cell walls).
- **thickness** of cell wall varies with cell function
- **primary cell wall** is outermost layer, composed of threadlike



Cellulose Microfibrils

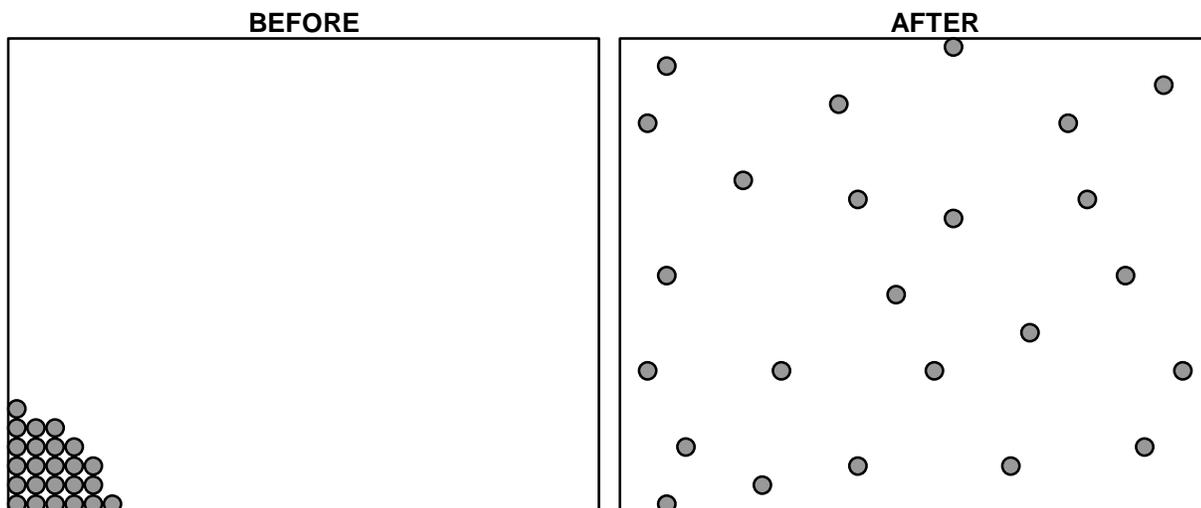
- **cellulose microfibrils**.
- sticky substance called **middle lamella** binds cells together

- woody plants also have a **secondary cell wall** which forms inside the primary wall. Composed of alternating layers of cellulose microfibrils reinforced with **lignin** (which adds strength). Function is **support** of large plants. **Wood** is made largely of secondary cell wall material.
- Cellulose of plant cell walls used by humans: **cotton, rayon, flax, hemp, paper, wood, paper** (paper has lignin removed to prevent yellowing). Lignin used in manufacture of rubber, plastics, pigments, adhesives.
- plant cell wall is **SEMI-PERMEABLE** (anything small enough to fit through openings in cellulose microfibrils will get through).
- plant cell therefore relies instead on its cell membrane to regulate what gets in and out.

**THERE ARE THREE GENERAL MEANS BY WHICH SUBSTANCE CAN ENTER AND EXIT CELLS:**

| Name  | Examples  |
|---|---|
| 1. <b>DIFFUSION</b>   | lipid-soluble molecules, water, gases                         |
| 2. <b>TRANSPORT BY CARRIERS</b> (i.e. active and facilitated transport)                 | sugars and amino acids<br>sugars, amino acids., ions          |
| 3. <b>ENDOCYTOSIS AND EXOCYTOSIS</b> (e.g. <b>pinocytosis</b> and <b>phagocytosis</b> ) | macromolecules (e.g. proteins), cells or subcellular material |

**Diffusion**



- diffusion is a **physical process** that can be observed with **any type of particle**. A **UNIVERSAL PHENOMENON**.
- Law of Diffusion:** particles **MOVE FROM THE AREA OF GREATER CONCENTRATION TO THE AREA OF LESSER CONCENTRATION UNTIL EQUALLY DISTRIBUTED**.
- for instance: *opening a perfume bottle* in corner of a room. The smell of perfume soon permeates the room because the molecules that make up the perfume have drifted to all parts of the room. e.g. dropping dye into water.
- movement by diffusion **requires no energy** to be added (although adding energy (i.e. heat) will speed it up).
- diffusion is a **slow** process. The rate of diffusion is affected by the **concentration gradient** (the **difference in concentration** of the diffusing molecules between the two regions), the **size & shape** of the molecules, and the **temperature**. Diffusion in liquid is slower than in gas. However, distribution of molecules in cytoplasm is **speeded up** by an ever-constant flow of the cytoplasm that is called **cytoplasmic streaming**.

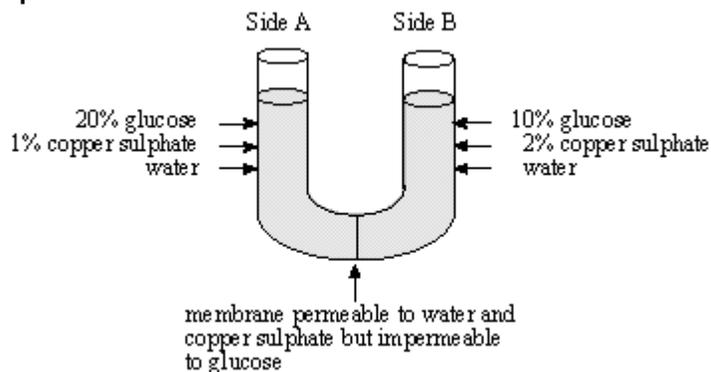
Three Ways of increasing the rate of diffusion:

1. **increase the temperature**
2. **increase the concentration gradient**
3. **decrease the size of the diffusing molecules**

- The properties of the cell membrane allow **few types** of molecules to pass by diffusion: **Lipid-soluble** molecules like **steroids** and **alcohols** can diffuse directly across because the membrane itself is made of lipids.
- **water** diffuses readily across membrane, probably through **charged, protein-lined pores** in the membrane (remember, water is not lipid-soluble) that will not allow anything else but water through.

## Osmosis

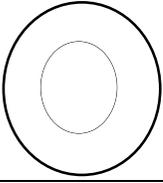
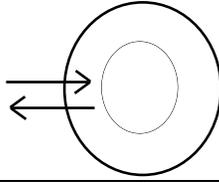
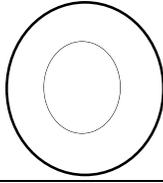
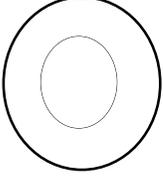
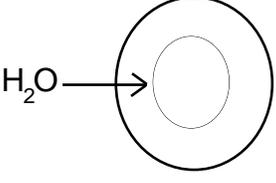
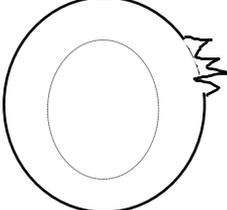
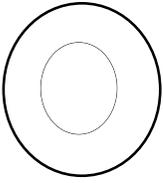
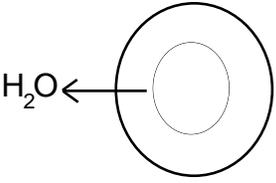
- the movement of water across a selectively permeable membrane is a **special case of diffusion** called **OSMOSIS**.
- **Definitions:**
  - Osmosis:** *the net movement of water molecules from the area of greater concentration to the area of lesser concentration across a selectively-permeable membrane.*
  - Solute:** particles which are dissolved in water
  - Solvent:** liquid which dissolves the solute. This is water when we are talking about osmosis.
  - Solution:** combination of solute and solvent.
  - Osmotic Pressure:** the pressure due to the flow of water from the area of greater concentration to the area of lesser concentration. The **greater the concentration difference** across the membrane, the **greater the osmotic pressure**.



*Explain what would happen to the concentrations of water, glucose, and copper sulphate on side A of this experiment.*

- In cellular systems, **water can move easily across membranes**, but other molecules can't. Therefore, it is often **only water** that can move and follow the law of diffusion. According to the law of diffusion, water will move from where it is more concentrated (i.e. solution that has **less** solute in it) to where it is less concentrated (i.e. solution that has **more** solute in it). This has important consequences on living systems.
- **Cells may be placed in solutions** that contain the same number of solute molecules per volume as the cell (= **isotonic solution**), a greater number of solute molecules per volume (= **hypertonic solution**), or a lesser number of solute molecules per volume than the cell (= **hypotonic solution**).

**Summary of what happens to ANIMAL CELLS placed in different tonicities of solution:**

| Tonicity of Solution Cell is Put Into  | Net Movement of Water   | Effect on Cell   |
|--|---|--|
| Isotonic<br>    | No <u>net</u> movement<br>   | Remains the same<br>                              |
| Hypotonic<br>   | Cell <b>gains</b> water<br>  | Cell <b>Swells &amp; May Burst</b><br><br>"lysis" |
| Hypertonic<br> | Cell <b>loses</b> water<br> | Cell <b>Shrinks</b><br><br>"crenation"           |

**Isotonic ("same" "strength") solution:**

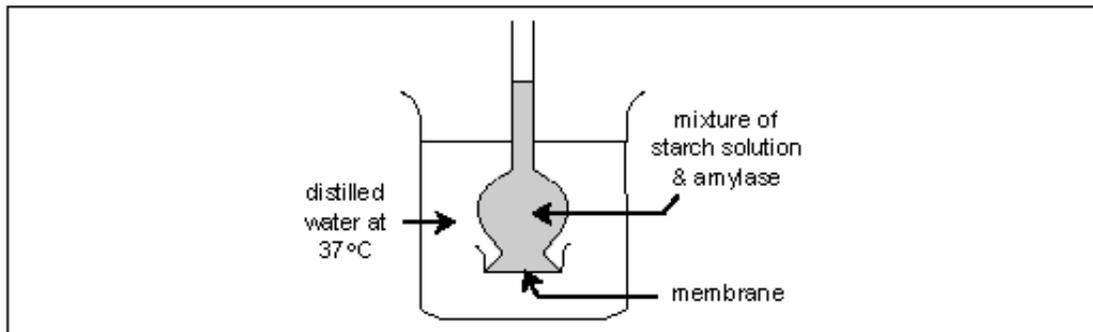
- **no net movement** of water across membrane.
- **Same number** of solute molecules per unit volume
- Cells placed in such a solution **neither** gain or lose water
- a **0.9 percent solution of NaCl is isotonic to red blood cells (RBC)**. *How can you tell this is so?*

**Hypotonic Solutions ("hypo" means "less than")**

- these solutions have **lower concentration** of **solute** than the cell contents.
- if cell placed in hypotonic solution, **water will enter cell**, it will **swell and possibly burst**.
- e.g. a salt solution with a concentration lower than 0.9% is hypotonic to RBC.

**Hypertonic Solutions ("greater" "strength")**

- **greater concentration** (symbol for concentration = "[ ]") of solute than the cell (and therefore a lesser [ ] of water)
- if a cell is placed in hypertonic solution, water will leave the cell and the cell will **shrive** up. This is called **CRENATION** in animal cells. e.g. a 10% solution of NaCl is hypertonic to RBC -- they'll shrink

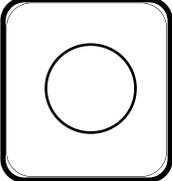
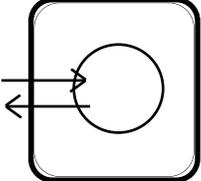
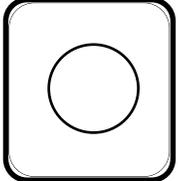
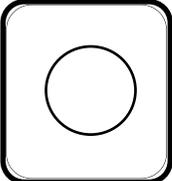
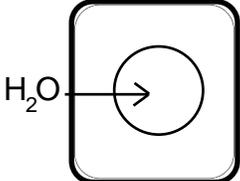
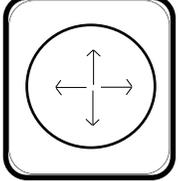
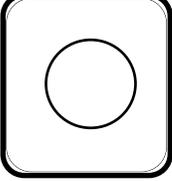
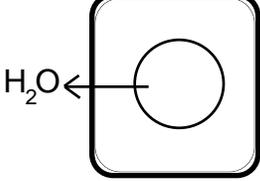


1. A student set up the experiment illustrated above and kept it at 37°C. After five minutes, the distilled water in the beaker was tested and found to contain a sugar but no starch.

a) What had occurred inside the tube? (1 mark)

### Significance of Tonicity to PLANT CELLS

Summary of what happens to PLANT CELLS placed in different tonicities of solution:

| Tonicity of Solution Cell is Put Into   | Net Movement of Water  | Effect on Cell   |
|---|--|--|
| Isotonic<br>    | No <u>net</u> movement<br>   | Remains the same<br>   |
| Hypotonic<br>  | Cell <b>gains</b> water<br> | Greater water pressure inside cell<br><br><b>"turgor pressure"</b>                |
| Hypertonic<br> | Cell <b>loses</b> water<br> | Cell Contents Shrink, but cell wall retains its shape<br><br><b>"plasmolysis"</b> |

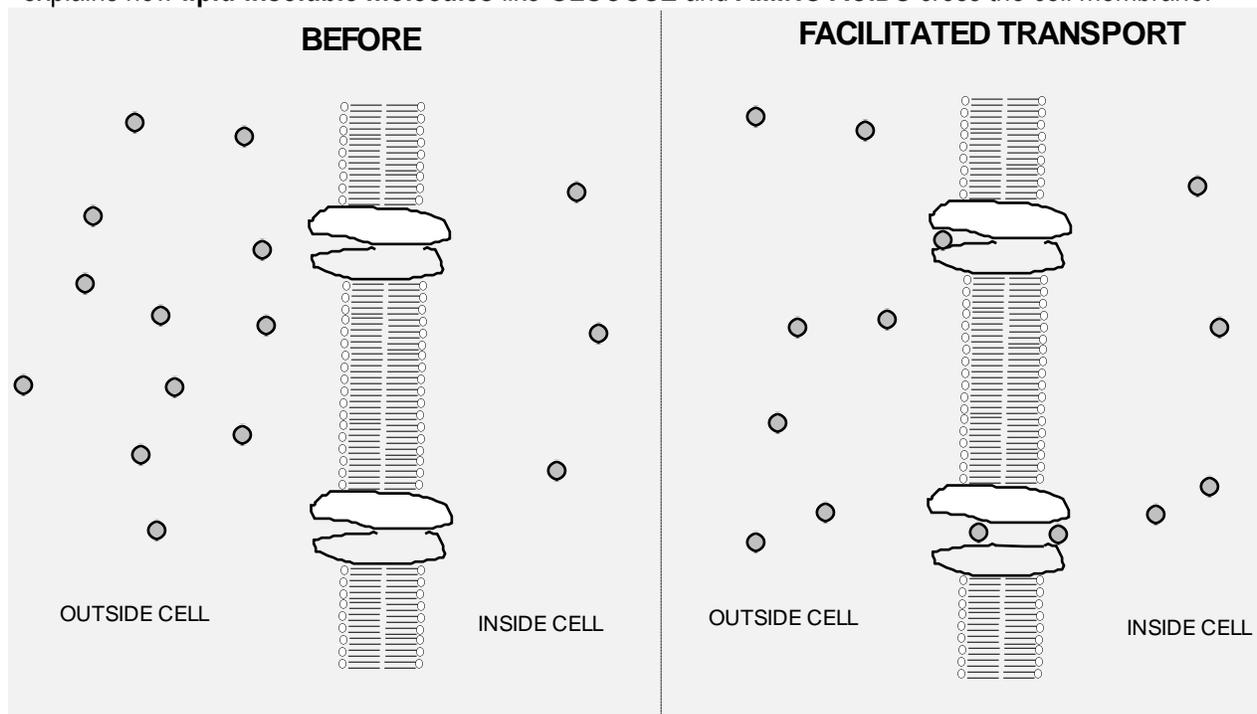
- **Hypertonic solutions** cause **PLASMOLYSIS** (shrinking of cell due to osmosis).
- **central vacuole loses water**, cell membrane shrinks and pulls away from cell wall.
- **Hypotonic solutions** causes **TURGOR PRESSURE**, against rigid cell wall (turgor pressure occurs when plant cells, placed in hypotonic solution, admit water. As water enters, **pressure builds up inside the cell** (*hydrostatic pressure*). When hydrostatic pressure = osmotic pressure, the plant is said to have developed turgor pressure).

- cell wall keeps cell from bursting
- **osmosis continues until turgor pressure = osmotic pressure**
- turgor pressure important for plant cells to **retain erect positions**. Now you should be able to explain why plants wilt when you don't water them!

## TRANSPORT BY CARRIERS

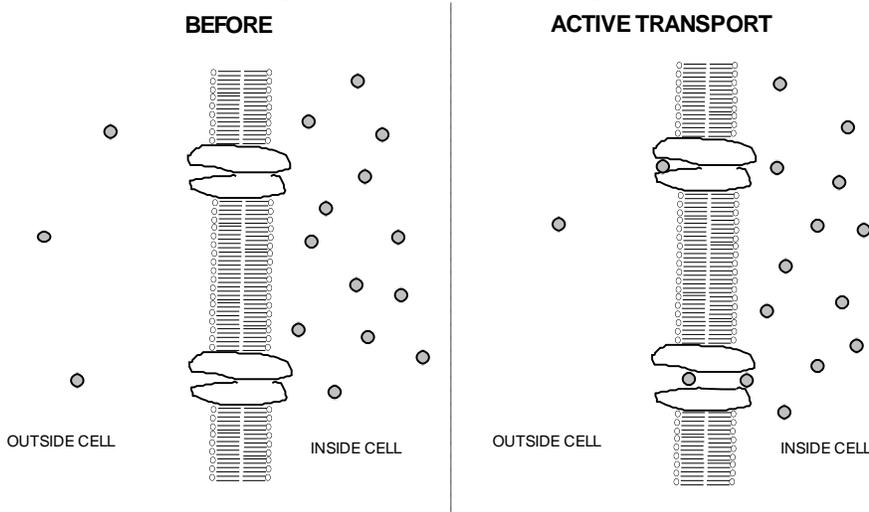
### **FACILITATED TRANSPORT**

- utilizes **PROTEIN CARRIERS** in cell membrane to control passage of molecules in and out of cell.
- are **highly specific** - each carrier passes only one type molecule
- molecules only pass along concentration gradient.
- **REQUIRES NO ENERGY** - is like diffusion in this sense
- explains how **lipid-insoluble molecules** like **GLUCOSE** and **AMINO ACIDS** cross the cell membrane.



## ACTIVE TRANSPORT

- also performed by **protein carriers**
- **REQUIRES ENERGY** (ATP)
- moves molecules **against the concentration gradient** (i.e. in the opposite direction of diffusion).



- molecules **move from area of lower concentration to area of higher concentration.**
- Active Transport **vitaly important** to organisms:  
e.g. **Iodine & Thyroid Gland.**  $[I^-]$  is low in blood, high in Thyroid Gland. Active Transport moves  $I^-$  from blood to thyroid.  
e.g.  $Na^+$  actively transported out of urine by kidney tubule cells  
e.g. **sodium/potassium pump** in nerve/muscle cells (see text). Moves  $Na^+$  from inside to

outside of cell, and  $K^+$  from outside to inside.

e.g.  $Na^+$   $Cl^-$  and **cystic fibrosis** - a **genetic disease**, usually **fatal**, caused by **blockage of  $Cl^-$  transport**.

## ENDOCYTOSIS AND EXOCYTOSIS

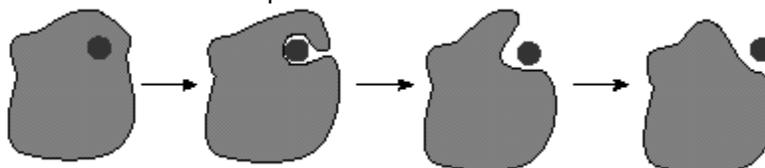
- another way to get molecules, **especially large particles**, in and out of cell.
- **ENDOCYTOSIS**: cell membrane forms a **vesicle** around the substance to be taken in.
  - **Phagocytosis**: what you call endocytosis if particles taken in really large (like other cells - e.g. human macrophages). Can be seen with **light microscope**.



- **Pinocytosis**: (= cell drinking) - same idea as phagocytosis, except smaller particles taken in (requires electron microscope to see).

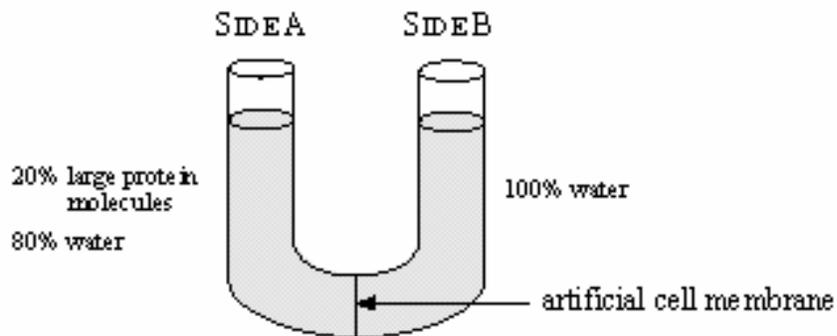


- **EXOCYTOSIS**: **Reverse of endocytosis**. Vacuole within cell fuses with cell membrane and the vacuole contents are deposited on the outside. Important in secretion and excretion in cells.



Multiple Choice Question:

What will happen to the protein solution in Side A of the apparatus in the diagram below?



- A. It will become less concentrated since water passes from **B** to **A**.
- B. It will become more concentrated since water passes from **B** to **A**.
- C. It will become more concentrated since water passes from **A** to **B**.
- D. It will become less concentrated since protein will move from **A** to **B**.